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Compounds: Writing Chemical
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Directions: Write the formulas of the following compounds.

1. ammonium phosphate

(NH₄)₃PO₄

_____ 2. iron (II) oxide

FeO_____ 3. iron (III) oxide

Fe₂O₃ _____ 4. carbon

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monoxide SKIP _____ 5.

calcium chloride CaCl₂

_____ 6. potassium nitrate

KNO₃ ...

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Formulas From Names 2CO 3 NH
4OH (NH 4) 2SO 4 (NH 4) 3PO
4 NH 4NO 3 K+!KCl K 2CO 3
KOH K 2SO 4 K 3PO 4 KNO 3
Ca²⁺!CaCl 2 ...

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Compounds. Terms in this set
(20) carbon monoxide. CO.
sodium chloride.

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sequence, roman numerals indicate the number of atoms of the element with which these are attached. as in carbon mono oxides there is one carbon and one...

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I need to write the formulas
for the names of the
following compounds: ammonium
phosphate iron (II) oxide
iron(III) oxide calcium
chloride potassium nitrate
magnesiumhydroxide aluminum

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Answer Key
sulfate copper (II) sulfate
lead (IV) chromate potassium
permanganate sodium hydrogen
carbonate zinc nitrate
aluminum sulfate sodium
hydroxide hydrogen sulfate i
really have absolutely no
idea at all how to do ...

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space, although there is A
formula used to indicate not
just the range of atoms.

Molecular formulas describe
the particular number and
variety of atoms within a
molecule of a compound. A
molecular formula isn't

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nitrate, ... with answers at the end of the page.. Once you find your worksheet(s), you can either click on the pop-out icon or download button to print or ...

Writing Chemical Formulas

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Practice Worksheets -
DSoftSchools

Ionic Bonding! and! Writing! For
Formulas! Name: !KEY!!! Part A.

Use the criss-cross method
to write the formulas
produced from the listed
ions. !Cl⁻!CO^{3 2-}!OH⁻!SO⁴

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2J!PO 4 3J!NO 3 J! Na+!NaCl
Na 2CO 3 NaOH Na 2SO 4 Na
3PO 4 NaNO 3 NH 4 +!NH 4Cl
(NH 4) 2CO 3 NH 4OH (NH 4)
2SO 4 (NH 4) 3PO 4 NH 4NO 3
K+!KCl K 2CO 3 KOH K 2SO 4 K
3PO 4 KNO 3 Ca2+!CaCl 2 CaCO
3 Ca(OH) 2 CaSO 4 Ca 3(PO 4)

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2 Ca(NO₃)₂

Ionic Bonding! and! Writing! Formulas! Name: !KEY!!! Part A.

Use ...

Write the formula for lead (IV) oxide; 1. Write the symbol and charge of the

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the negative charge to become the subscript of the cation. 3.

5.5: Writing Formulas for Ionic Compounds - Chemistry

...

- Name these 1) FeCl_3 2) Ag

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3) AsO_4^{3-} 4) Ba(OH)_2 5) H_2O

6) H_3PO_3 •Write the formulas for these compounds

7) aluminum iodide

8) ammonium phosphate 9) nitric acid

10) hydrogen sulfide •Separate the ions, give their names and charges and

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name the compound. Change
the anion into the acid and
name that acid . 10) MgCO_3

Naming and Writing Formulas
for Acids!

Chemical Formula Writing
Worksheet Solutions. Write

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Answer Key
Chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the

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“chloride” anion, so you should write “ZnCl₂”, as shown.

Ionic Compound Formula
Writing Worksheet

Writing names name first
element in formula- do not

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Answer Key
change name name second
element and change ending to
-ide use prefixes to tell
how many of each element no
mono- on first element
Writing Formulas use
prefixes to determine what
subscripts should be Based

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on your rules, name these
compounds. 1. OF 2

Compound Names and Formulas
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Thorson's Science 10 Blog |

A fine ...

Answer to Write the names of
the following compounds:

Na₂SO₄ Mg₃N₂ N₂O₅ SF₆ CO

Write the formulas of the
following compounds: pota...

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periodic table, and chemical
bonding. These topics are
followed by discussions on
writing names and formulas
of compounds; chemical
formulas and the mole;
chemical reactions;

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changes and the adoption of IUPAC periodic table conventions are highlights of the narrative revisions, along with changes to the discussion of d orbitals. In-text examples have been reformatted to facilitate

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